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# Yan-Min Liu, Chun-Yan Liu\* and Ai-Guo Meng

Department of Pharmaceuticals, North China Coal Medical Colledge, Tangshan 063000, People's Republic of China

Correspondence e-mail: chyliu2004@126.com

#### **Key indicators**

Single-crystal X-ray study T = 294 K Mean  $\sigma$ (C–C) = 0.010 Å R factor = 0.056 wR factor = 0.157 Data-to-parameter ratio = 14.6

For details of how these key indicators were automatically derived from the article, see http://journals.iucr.org/e.

# 1,1'-Ethylenebis(4-aminopyridinium) dibromide

In the title compound,  $C_{12}H_{16}N_4^{2+}\cdot 2Br^-$ , the molecule has a centre of symmetry at the mid-point of the central C–C bond and a mirror plane passes through the amino N atom, the attached C atom, the ring N atom, and the linking chain C atom; the Br anion lies on a position of site symmetry *m*. The two pyridyl rings are parallel to each other.

#### Comment

N-(2-Bromoethyl)pyridinium cations and 1,1'-ethylenebispyridinium dications are potential sources of N-vinylpyridinium cations *via* base-catalysed elimination reactions (Katritzky & Rubio, 1983). Bunting *et al.*, (1992) have conducted research on the elimination reaction of 1,1'-ethylenebis(4-aminopyridinium) dibromide, (I), in an aqueous base medium. Here we report the crystal structure of (I).



The molecular structure of (I) is shown in Fig. 1. The molecule has a centre of symmetry at the mid-point of the C6– C6(1 – x, –y, 1 – z) bond and a mirror plane passes through atoms C1, N2 and C6. Atom Br1 lies on a position of site symmetry *m*. The two pyridyl rings are parallel to each other. The bond N1–C1 is short enough [1.325 (12) Å] to indicate significant double-bond character. The planar geometry around the amino N atom suggests strong conjugation with the  $\pi$ -system of the pyridyl ring. The N<sup>+</sup>···N<sup>+</sup> distance in (I) is 3.740 (1) Å, similar to the value previously reported (*ca* 3.75 Å) in the 1,2-bis(pyridinium)ethane dication (Loeb & Wisner, 1998). The bond lengths and angles are unexceptional. The molecular structure is stabilized by an intermolecular N– H···Br hydrogen bond (Table 1).



View of the structure of the cation of (I), showing the atom-labelling scheme. Displacement ellipsoids are drawn at the 30% probability level and H atoms are shown as small spheres of arbitrary radii. The bromide ions have been omitted for clarity. [Symmetry codes: (A) x, -y, z; (B) 1 - x. -y, 1 - z; (C) 1 - x, y, 1 - z.]

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# Experimental

1,1'-Ethylenebis(4-aminopyridinium) dibromide was prepared according to the method of Bunting (1992). Single crystals suitable for analysis were grown by evaporation of an ethanol solution.

#### Crystal data

 $C_{12}H_{16}N_4^{2+}\cdot 2Br^{-}$   $M_r = 376.11$ Monoclinic, C2/m a = 13.121 (4) Å b = 6.4876 (19) Å c = 8.406 (2) Å  $\beta = 104.023$  (5)° V = 694.2 (4) Å<sup>3</sup> Z = 2

### Data collection

Bruker SMART 1000 CCD areadetector diffractometer  $\varphi$  and  $\omega$  scans Absorption correction: multi-scan (*SADABS*; Sheldrick, 1996)  $T_{\min} = 0.214, T_{\max} = 0.350$ 1850 measured reflections

#### Refinement

refinement

Refinement on  $F^2$   $R[F^2 > 2\sigma(F^2)] = 0.056$   $wR(F^2) = 0.157$  S = 1.11758 reflections 52 parameters H atoms treated by a mixture of independent and constrained Cell parameters from 1061 reflections  $\theta = 3.2-26.2^{\circ}$  $\mu = 5.83 \text{ mm}^{-1}$ T = 294 (2) K Block, colorless  $0.32 \times 0.22 \times 0.18 \text{ mm}$ 

 $D_r = 1.799 \text{ Mg m}^{-3}$ 

Mo  $K\alpha$  radiation

758 independent reflections 638 reflections with  $I > 2\sigma(I)$   $R_{int} = 0.033$   $\theta_{max} = 26.2^{\circ}$   $h = -16 \rightarrow 14$   $k = -8 \rightarrow 7$  $l = -10 \rightarrow 7$ 

+ 6.17 <i>P</i> ] where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 1.52 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.93 \text{ e } \text{\AA}^{-3}$	$w = 1/[\sigma^2(F_o^2) + (0.0778P)^2]$
where $P = (F_o^2 + 2F_c^2)/3$ $(\Delta/\sigma)_{max} < 0.001$ $\Delta\rho_{max} = 1.52 \text{ e } \text{\AA}^{-3}$ $\Delta\rho_{min} = -0.93 \text{ e } \text{\AA}^{-3}$	+ 6.17P]
$\begin{array}{l} (\Delta/\sigma)_{\rm max} < 0.001 \\ \Delta\rho_{\rm max} = 1.52 \ {\rm e} \ {\rm \AA}^{-3} \\ \Delta\rho_{\rm min} = -0.93 \ {\rm e} \ {\rm \AA}^{-3} \end{array}$	where $P = (F_0^2 + 2F_c^2)/3$
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	$\Delta \rho_{\rm min} = -0.93 \ {\rm e} \ {\rm \AA}^{-3}$

## Table 1

Hydrogen-bond geometry (Å,  $^{\circ}$ ).

$D - H \cdot \cdot \cdot A$	D-H	$H \cdot \cdot \cdot A$	$D \cdots A$	$D - \mathbf{H} \cdots A$
$N1-H1A\cdots Br1^{i}$	0.83 (8)	2.70 (8)	3.516 (3)	166 (7)
Symmetry code: (i) -	$x + \frac{1}{2}, -y - \frac{1}{2}, -z$	ζ.		

Atom H1A was located in a difference Fourier map and refined freely; the final N1–H1A bond length is 0.83 (8) Å; the  $U_{iso}(H)$  value was set equal to  $1.5U_{eq}(N)$ . Other H atoms were positioned geometrically and constrained to ride on their parent atoms [C–H = 0.93 Å for aromatic and 0.97 Å for CH<sub>2</sub> groups, with  $U_{iso}(H) = 1.2U_{eq}(C)$ ]. The maximum residual electron-density peak is located 0.87 Å from atom Br1.

Data collection: *SMART* (Bruker, 1997); cell refinement: *SAINT* (Bruker, 1997); data reduction: *SAINT*; program(s) used to solve structure: *SHELXS97* (Sheldrick, 1997); program(s) used to refine structure: *SHELXL97* (Sheldrick, 1997); molecular graphics: *SHELXTL* (Bruker, 1997); software used to prepare material for publication: *SHELXTL*.

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